

# Course Syllabus



## Organic Chemistry II

### CHEM 332, Summer 2024

#### Online Only


**Instructor:** Prof. Levi Stanley (3101E Hach Hall, Phone: 515-294-3609, e-mail: [lstanley@iastate.edu](mailto:lstanley@iastate.edu) (<mailto:lstanley@iastate.edu>))- please include 332 in the subject of messages)

**Office hours:** TWThF 9-10 AM or by appointment in person and over Cisco Webex.

**Communication:** Email – I welcome your email with questions about the course. That said, I receive a large amount of email each day. To ensure your email is answered, please follow these rules: 1) always include “332” somewhere in the subject line and 2) always send your email from an iastate.edu address.

**Textbook:** *Organic Chemistry*, 4<sup>th</sup> Edition, by David Klein. An optional *Student Study Guide & Solutions Manual for Organic Chemistry*, 4<sup>th</sup> Edition is also available and is recommended. Students should consider buying a molecular model kit if you have trouble visualizing in 3-dimensions. Many students find *Pushing Electrons: A Guide for Students of Organic Chemistry, 4th Ed.* by Daniel Weeks and Art Winter to be a very helpful supplement.

**Canvas:** Course information including a course syllabus, lectures, previous exams, current exams, supplemental materials, announcements, and grades will be uploaded to Canvas.

**Homework:** We will use the Wiley online homework system in this course ([wileyplus.com](http://wileyplus.com)  (<http://wileyplus.com/>)). WileyPLUS will be integrated into Canvas. You should not need to leave Canvas to complete your homework. Although the online homework is the only graded homework, prior student performance suggests that working end-of-chapter problems from your text greatly improves understanding. **You are strongly encouraged to work as many end-of-chapter problems as possible!**

**Grading:** The course will be graded out of 400 points. The point breakdown is as follows: 100 points from online homework and 300 points from midterm exams. Your final grade may be curved based on the performance of the class, but you are guaranteed the following grades: >85% for an A-, >75% for a B-, >60% for a C-, >50% for a D-.

**Exams:** There will be four 100-point midterm exams. The lowest midterm exam score from the first three midterm exams will be dropped. The exams are open-note and open-book exams. You may not work with other people on your exams. **You may not use Chegg or related sites to complete your exams.** Exams will open on Canvas at 9 AM on the scheduled exam date and close at 5 PM on the scheduled exam date. You will be required to upload a scanned copy of your exam to Canvas by 5 PM on the exam date. **All exams are cumulative.** At my discretion, in-class exam scores may be adjusted upward to balance the difficulty of different in-class midterm exams. Cheating on an exam will earn a grade of zero for that exam. CHEM 332 exams from previous semesters are available in Canvas. Exams will be given only on the dates listed in the syllabus. **There will be no make-up exams.**

**Re-grade Requests:** Grading of exams is not always perfect, and I believe you should receive the grade you deserve. Students may request an exam be re-graded within one week after the graded exams are returned. To request an exam re-grade, please write on the front of the exam to describe the nature of your re-grade request.

**Course Content:** We will cover chapters 12-13, chapters 16-22, and chapters 24 in *Organic Chemistry*, 4<sup>th</sup> Edition, by David Klein. These chapters and the supplemental sections build upon themselves and represent a large amount of material. Thus, it is extremely important that you stay current with your reading and homework assignments. The material will be covered in the order listed in the syllabus. However, the timing may vary based upon student understanding of the material.

**Learning Objectives:** Organic Chemistry II will be a demanding course. You will be asked to digest a large amount of material in a relatively short time. In addition, you will be asked to master the basic language and fundamental concepts (such as reactivity trends, steric and electronic effects, and basic kinetic and thermodynamic effects) that are the required foundation to solve problems in organic chemistry. The sheer volume of information to be covered is such that rote memorization becomes impractical as a singular learning strategy. Thus, it is critical that you are able to connect to and build upon new and existing knowledge of fundamental principles and concepts in organic chemistry. The most successful students in organic chemistry are often those who understand basic reactivity

(nucleophile or electrophile, acid or base) of common functional groups and reagents and are able to apply trends to new problems.

At the end of Chem 332, you will have been through as much organic chemistry as most beginning Ph.D. students in chemistry! That said, most of you have long-term learning and career goals in which chemistry is just a part. The following is a summary of what we want you to know or be able to do upon completion of the year sequence of organic chemistry.

1. You should have a good understanding of molecular structure. This includes sigma and pi bonding, strain, aromaticity, and stereochemistry. You should have a good grasp of three-dimensional structure of molecules and should understand that intermolecular interactions depend on structural relationships.
2. You should be able to recognize the reactive parts of molecules. In particular, you should be able to identify nucleophilic and electrophilic centers. You should understand how to make certain centers more nucleophilic or electrophilic, e.g., by deprotonating or protonating them.
3. You should be able to recognize types of reactions that you see in different contexts. For example, you should recognize substitutions, additions, oxidations, and reductions, no matter whether you see them in organic chemistry or in a biochemistry class.
4. You will have a good fundamental understanding on the chemical reactivity of several common classes of reactive intermediates, especially cationic and anionic centers.
5. You should be able to understand complex reaction mechanisms if they are presented to you. You should be able to suggest reasonable reaction mechanisms for almost every reaction you know, and – based on knowing something about the reactive parts of molecules and reactive intermediates – make a good guess about the mechanism of a new reaction presented to you.
6. You should have a reasonable repertoire of reactions that you can call upon to imagine how to synthesize a molecule of modest complexity or follow its synthesis or biosynthesis as presented to you. In practice, many of you will find this most useful in biological and biochemical contexts.
7. From Chemistry 332 in particular, you should develop a thorough understanding of carbonyl chemistry, including addition, reduction, and oxidation. You will learn several ways in which carbonyl compounds can be used to construct new carbon-carbon bonds. Yes, there are many other functionalities covered in 332, but this is the heart of the reactivity we examine.

**Learning Environment:** My goal for this course is to create an active learning environment. To accomplish this goal, you as students must be willing to engage in the learning process. This goal will be a bit more challenging in an online format, but I encourage you to bring questions to office hours each day.

**Free Expression:** Iowa State University supports and upholds the First Amendment protection of [freedom of speech \(https://freespeech.iastate.edu/\)](https://freespeech.iastate.edu/) and the principle of [academic freedom \(https://www.iowaregents.edu/plans-and-policies/board-policy-manual/310-academic-freedom\)](https://www.iowaregents.edu/plans-and-policies/board-policy-manual/310-academic-freedom) in order to foster a learning environment where open inquiry and the vigorous debate of a diversity of ideas are encouraged. Students will not be penalized for the content or viewpoints of their speech as long as student expression in a class context is germane to the subject matter of the class and conveyed in an appropriate manner.

**Academic Dishonesty:** The class will follow Iowa State University's policy on academic misconduct ([5.1 in the Student Code of Conduct \(https://www.policy.iastate.edu/policy/SDR\)](https://www.policy.iastate.edu/policy/SDR)). Students are responsible for adhering to university policy and the expectations in the course syllabus and on coursework and exams, and for following directions given by faculty, instructors, and Testing Center regulations related to coursework, assessments, and exams. Anyone suspected of academic misconduct will be reported to the [Office of Student Conduct in the Dean of Students Office \(https://www.studentconduct.dso.iastate.edu/academic-misconduct/armfacultystaff\)](https://www.studentconduct.dso.iastate.edu/academic-misconduct/armfacultystaff). Information about academic integrity and the value of completing academic work honestly can be found in the [Iowa State University Academic Integrity Tutorial. \(https://iastate.pressbooks.pub/academicintegrity/\)](https://iastate.pressbooks.pub/academicintegrity/)

**Accessibility:** Iowa State University is committed to advancing equity, access, and inclusion for students with disabilities. Promoting these values entails providing reasonable accommodations where barriers exist to students' full participation in higher education. Students in need of accommodations or who experience accessibility-related barriers to learning should work with Student Accessibility Services (SAS) to identify resources and support available to them. Staff at SAS collaborate with students and campus partners to coordinate accommodations and to further the academic excellence of students with disabilities. Information about SAS is available online at [www.sas.dso.iastate.edu \(http://www.sas.dso.iastate.edu\)](http://www.sas.dso.iastate.edu), by email at [accessibility@iastate.edu](mailto:accessibility@iastate.edu), or by phone at 515-294-7220.

**Discrimination and Harassment:** Iowa State University does not discriminate on the basis of race, color, age, ethnicity, religion, national origin, pregnancy, sexual orientation, gender identity, genetic information, sex, marital status, disability, or status as a U.S. Veteran. Inquiries regarding non-discrimination policies may be directed to Office of Equal Opportunity, 3410 Beardshear Hall, 515 Morrill Road, Ames, Iowa 50011, Tel. 515-294-7612, Hotline 515-294-1222, email [eooffice@iastate.edu \(mailto:eooffice@iastate.edu\)](mailto:eooffice@iastate.edu).

**Mental Health and Well-Being Resources:** At Iowa State, we're committed to your success and well-being. As a Cyclone, you can access 24/7 resources, services, and people dedicated to helping you

achieve your goals and be your best in and out of the classroom. Whether you need academic support or just someone to talk to, we're here for you at Cyclone Support ([cyclonesupport.iastate.edu](http://cyclonesupport.iastate.edu)).

**Prep Week:** This class follows the Iowa State University Prep Week policy as noted the ISU Policy Library; as well as section 10.6.4 of the Faculty Handbook. Visit the [ISU Policy Library website \(https://www.policy.iastate.edu/\)](https://www.policy.iastate.edu/) (<http://www.policy.iastate.edu/>) for policy wording.

**Religious Accommodation:** Iowa State University welcomes diversity of religious beliefs and practices, recognizing the contributions differing experiences and viewpoints can bring to the community. There may be times when an academic requirement conflicts with religious observances and practices. If that happens, students may request reasonable accommodation for religious practices. In all cases, you must put your request in writing. The instructor will review the situation in an effort to provide a reasonable accommodation when possible to do so without fundamentally altering a course. For students, you should first discuss the conflict and your requested accommodation with your professor at the earliest possible time. You or your instructor may also seek assistance from the [Dean of Students Office website \(https://dso.iastate.edu/\)](https://dso.iastate.edu/) (<http://dso.iastate.edu/>) or via phone 515-294-1020 or the [Office of Equal Opportunity website \(https://www.eoc.iastate.edu/\)](https://www.eoc.iastate.edu/) (<https://www.eoc.iastate.edu/>) or via phone 515-294-7612.

**Student Code of Conduct:** Students in this course are responsible for being familiar with the Iowa State University [Student Code of Conduct \(https://www.policy.iastate.edu/policy/SDR\)](https://www.policy.iastate.edu/policy/SDR) in the ISU Policy Library.

## Schedule

Date	Chapter	Topic
5/13/24	12	Alcohols and Phenols
5/14/24	12	Alcohols and Phenols
5/15/24	12	Alcohols and Phenols
5/16/24	13	Ethers and Epoxides; Thiols and Sulfides
5/17/24	13	Ethers and Epoxides; Thiols and Sulfides

5/20/24	13	Ethers and Epoxides; Thiols and Sulfides
5/21/24	13	Ethers and Epoxides; Thiols and Sulfides
5/22/24	<b>Exam 1 (Chapters 12 and 13)</b>	
5/23/24	16	Conjugated Pi Systems and Pericyclic Reactions
5/24/24	16	Conjugated Pi Systems and Pericyclic Reactions
5/27/24	No Class, University Holiday	
5/28/24	16	Conjugated Pi Systems and Pericyclic Reactions
5/29/24	17	Aromatic Compounds
5/30/24	17	Aromatic Compounds
5/31/24	17	Aromatic Compounds
6/3/24	18	Aromatic Substitution Reactions
6/4/24	18	Aromatic Substitution Reactions
6/5/24	18	Aromatic Substitution Reactions
6/6/24	18	Aromatic Substitution Reactions
6/7/24	19	Aldehydes and Ketones
6/10/24	<b>Exam 2 (Chapters 17, 18, and 19)</b>	
6/11/24	19	Aldehydes and Ketones
6/12/24	19	Aldehydes and Ketones

6/13/24	20	Carboxylic Acids and Their Derivatives
6/14/24	20	Carboxylic Acids and Their Derivatives
6/17/24	20	Carboxylic Acids and Their Derivatives
6/18/24	20	Carboxylic Acids and Their Derivatives
6/19/24	<b>Exam 3 (Chapters 19 and 20)</b>	
6/20/24	21	Alpha Carbon Chemistry: Enols and Enolates
6/21/24	21	Alpha Carbon Chemistry: Enols and Enolates
6/24/24	21	Alpha Carbon Chemistry: Enols and Enolates
6/25/24	21	Alpha Carbon Chemistry: Enols and Enolates
6/26/24	21	Alpha Carbon Chemistry: Enols and Enolates
6/27/24	22	Amines
6/28/24	22	Amines
7/1/24	22	Amines
7/2/24	<b>Exam 4 (Chapters 21 and 22)</b>	
7/3/24	24	Carbohydrates
7/4/23	No Class, University Holiday	
7/5/24	24	Carbohydrates