

Organic Chemistry II
CHEM 332, Fall 2023

Monday, Wednesday, and Friday 2:15 to 3:05 PM, 1411 Lagomarcino Hall

Instructor: Prof. Levi Stanley (3101E Hach Hall, Phone: 515-294-3609, e-mail: lstanley@iastate.edu - please include 332 in the subject of messages)

Office hours: MWTh 1:10-2 PM or by appointment.

Communication: Email – I welcome your email with questions about the course. That said, I receive a large amount of email throughout the semester. To ensure your email is answered, please follow these rules: 1) always include “332” somewhere in the subject line; 2) always send your email from an iastate.edu address; and 3) DO NOT use Canvas to send me emails.

Textbook: *Organic Chemistry*, 4th Edition, by David Klein. An optional *Student Study Guide & Solutions Manual for Organic Chemistry*, 4th Edition is also available and is recommended. Students should consider buying a molecular model kit if you have trouble visualizing in 3-dimensions. Many students find *Pushing Electrons: A Guide for Students of Organic Chemistry, 4th Ed.* by Daniel Weeks and Art Winter to be a very helpful supplement.

Canvas: Course information including a course syllabus, lectures, previous exams, current exams, supplemental materials, announcements, and grades will be uploaded to Canvas. Please do not use Canvas to contact me. Please use my university email address to contact me.

Homework: We will use the Wiley online homework system in this course (wileyplus.com). WileyPLUS will be integrated into Canvas. You should not need to leave Canvas to complete your homework. Although the online homework is the only graded homework, prior student performance suggests that working end-of-chapter problems from your text greatly improves understanding. **You are strongly encouraged to work as many end-of-chapter problems as possible!**

Grading: The course will be graded out of 600 points. The point breakdown is as follows: 100 points from online homework, 300 points from midterm exams, and 200 points from the final exam. Your final grade may be curved based on the performance of the class, but you are guaranteed the following grades: >85% for an A-, >75% for a B-, >60% for a C-, >50% for a D-.

Exams: There will be four 100-point midterm exams. The exams will be held in person during normal class times as scheduled on the syllabus. **All exams are cumulative.** The best three exam scores from your midterm exams plus the final exam score will be used in the calculation of your final grade. At my discretion, in-class exam scores may be adjusted upward to balance the difficulty of different in-class midterm exams. Cheating on an exam will earn a grade of zero for that exam. Such an exam score cannot be dropped. CHEM 332 exams from previous semesters are available in Canvas. Exams will be given only on the dates listed in the syllabus. **There will be no make-up exams.**

Re-grade Requests: Grading of exams is not always perfect, and I believe you should receive the grade you earn. Students may request an exam be re-graded within one week after the graded exams are returned. To request an exam re-grade, please write on the front of the exam to describe the nature of your re-grade request.

Course Content: We will cover chapters 12-13, chapters 16-22, and chapters 24-25 in *Organic Chemistry*, 4th Edition, by David Klein. These chapters and the supplemental sections build upon themselves and represent a large amount of material. Thus, it is extremely important that you stay current with your reading and homework assignments. The material will be covered in the order listed in the syllabus. However, the timing may vary based upon student understanding of the material.

Learning Objectives: Organic Chemistry II will be a demanding course. You will be asked to digest a large amount of material in a relatively short time. In addition, you will be asked to master the basic language and fundamental concepts (such as reactivity trends, steric and electronic effects, and basic kinetic and thermodynamic effects) that are the required foundation to solve problems in organic chemistry. The sheer volume of information to be covered is such that rote memorization becomes impractical as a singular learning strategy. Thus, it is critical that you are able to connect to and build upon new and existing knowledge of fundamental principles and concepts in organic chemistry. The most successful students in organic chemistry are often those who understand basic reactivity (nucleophile or electrophile, acid or base) of common functional groups and reagents and are able to apply trends to new problems.

At the end of Chem 332, you will have been through as much organic chemistry as most beginning Ph.D. students in chemistry! That said, most of you have long-term learning and career goals in which chemistry is just a part. The following is a summary of what we want you to know or be able to do upon completion of the year sequence of organic chemistry.

1. You should have a good understanding of molecular structure. This includes sigma and pi bonding, strain, aromaticity, and stereochemistry. You should have a good grasp of three-dimensional structure of molecules and should understand that intermolecular interactions depend on structural relationships.
2. You should be able to recognize the reactive parts of molecules. In particular, you should be able to identify nucleophilic and electrophilic centers. You should understand how to make certain centers more nucleophilic or electrophilic, e.g., by deprotonating or protonating them.
3. You should be able to recognize types of reactions that you see in different contexts. For example, you should recognize substitutions, additions, oxidations, and reductions, no matter whether you see them in organic chemistry or in a biochemistry class.
4. You will have a good fundamental understanding on the chemical reactivity of several common classes of reactive intermediates, especially cationic and anionic centers.
5. You should be able to understand complex reaction mechanisms if they are presented to you. You should be able to suggest reasonable reaction mechanisms for almost every reaction you know, and – based on knowing something about the reactive parts of molecules and reactive intermediates – make a good guess about the mechanism of a new reaction presented to you.
6. You should have a reasonable repertoire of reactions that you can call upon to imagine how to synthesize a molecule of modest complexity or follow its synthesis or biosynthesis as presented to you. In practice, many of you will find this most useful in biological and biochemical contexts.
7. From Chemistry 332 in particular, you should develop a thorough understanding of carbonyl chemistry, including addition, reduction, and oxidation. You will learn several ways in which carbonyl compounds can be used to construct new carbon-carbon bonds. Yes, there are many other functionalities covered in 332, but this is the heart of the reactivity we examine.

Learning Environment: My goal for this course is to create an active learning environment. To accomplish this goal, you as students must be willing to engage in the learning process. This goal will be a bit more challenging in an online format, but I encourage you to bring questions to office hours each day.

Academic Integrity

Academic Integrity, based on the values of honesty, trust, fairness, respect, and responsibility, is a fundamental principle of scholarship in higher education. Iowa State's Academic and Research Misconduct Policy prohibits: plagiarism (using another person's writing or copying any work without proper citation); falsification; unauthorized collaboration during a test or on an assignment or substitution for another student to take an exam, course or test; and other forms of academic dishonesty. If you are to benefit from this class and be properly evaluated for your contributions, it is important for you to be familiar with and follow Iowa [State's Academic Dishonesty and Research Misconduct policies](#). Students are also encouraged to review these [Student Resources](#). Work that violates this policy will not be tolerated. Students who are found responsible for a violation of the Academic Misconduct Policy will have both a university process sanction and an academic outcome that could include a failing grade on the assignment or exam, or a failing grade for the course.

Accessibility

Iowa State University is committed to assuring that all educational activities are free from discrimination and harassment based on disability status. Students requesting accommodations for a documented disability are required to work directly with staff in Student Accessibility Services (SAS) to establish eligibility and learn about related processes before accommodations will be identified. After eligibility is established, SAS staff will create and issue a Notification Letter for each course listing approved reasonable accommodations. This document will be made available to the student and instructor either electronically or in hard-copy every semester. Students and instructors are encouraged to review contents of the Notification Letters as early in the semester as possible to identify a specific, timely plan to deliver/receive the indicated accommodations. Reasonable accommodations are not retroactive in nature and are not intended to be an unfair advantage. Additional information or assistance is available online at www.sas.dso.iastate.edu, by contacting SAS staff by email at accessibility@iastate.edu, or by calling 515-294-7220. Student Accessibility Services is a unit in the Dean of Students Office located at 1076 Student Services Building.

Discrimination and Harassment

Iowa State University does not discriminate on the basis of race, color, age, ethnicity, religion, national origin, pregnancy, sexual orientation, gender identity, genetic information, sex, marital status, disability, or status as a U.S. Veteran. Inquiries regarding non-discrimination policies may be directed to Office of Equal Opportunity, 3410 Beardshear Hall, 515 Morrill Road, Ames, Iowa 50011, Tel. 515-294-7612, Hotline 515-294-1222, email eooffice@iastate.edu.

Free Expression

Iowa State University supports and upholds the First Amendment protection of [freedom of speech](#) and the principle of [academic freedom](#) in order to foster a learning environment where open inquiry and the vigorous debate of a diversity of ideas are encouraged. Students will not be penalized for the content or viewpoints of their speech as long as student expression in a class context is germane to the subject matter of the class and conveyed in an appropriate manner.

Prep Week

This class follows the Iowa State University Prep Week policy, as noted in the ISU Policy Library and section 10.6.4 of the Faculty Handbook. Visit the [ISU Policy Library website](#) for policy wording.

Religious Accommodations

Iowa State University welcomes diversity of religious beliefs and practices, recognizing the contributions differing experiences and viewpoints can bring to the community. There may be times when an academic requirement conflicts with religious observances and practices. If that happens, students may request reasonable accommodation for religious practices. In all cases, you must put your request in writing. The instructor will review the situation in an effort to provide a reasonable accommodation when possible to do so without fundamentally altering a course. For students, you should first discuss the conflict and your requested accommodation with your professor at the earliest possible time. You or your instructor may also seek assistance from the [Dean of Students Office](#) at 515-294-1020 or [the Office of Equal Opportunity](#) at 515-294-7612.

Contact Information for Academic Issues: If you are experiencing, or have experienced, a problem with any of the above issues, email academicissues@iastate.edu

Schedule

Date	Chapter	Topic
8/21/23		Course Orientation and Introduction
8/23/23	12	Alcohols and Phenols
8/25/23	12	Alcohols and Phenols
8/28/23	12	Alcohols and Phenols
8/30/23	13	Ethers and Epoxides; Thiols and Sulfides
9/1/23	13	Ethers and Epoxides; Thiols and Sulfides
9/4/23		No Class, University Holiday
9/6/23	13	Ethers and Epoxides; Thiols and Sulfides
9/8/23	13	Ethers and Epoxides; Thiols and Sulfides
9/11/23		Exam 1 (Chapters 12 and 13)
9/13/23	16	Conjugated Pi Systems and Pericyclic Reactions
9/15/23	16	Conjugated Pi Systems and Pericyclic Reactions
9/18/23	16	Conjugated Pi Systems and Pericyclic Reactions
9/20/23	17	Aromatic Compounds
9/22/23	17	Aromatic Compounds
9/25/23	17	Aromatic Compounds
9/27/23	18	Aromatic Substitution Reactions
9/29/23	18	Aromatic Substitution Reactions
10/2/23	18	Aromatic Substitution Reactions
10/4/23	18	Aromatic Substitution Reactions
10/6/23		Exam 2 (Chapters 16, 17, and 18)
10/09/23	19	Aldehydes and Ketones
10/11/23	19	Aldehydes and Ketones
10/13/23	19	Aldehydes and Ketones
10/16/23	20	Carboxylic Acids and Their Derivatives
10/18/23	20	Carboxylic Acids and Their Derivatives
10/20/23	20	Carboxylic Acids and Their Derivatives
10/23/23	20	Carboxylic Acids and Their Derivatives
10/25/23		Exam 3 (Chapters 19 and 20)
10/27/23	21	Alpha Carbon Chemistry: Enols and Enolates
10/30/23	21	Alpha Carbon Chemistry: Enols and Enolates
11/1/23	21	Alpha Carbon Chemistry: Enols and Enolates
11/3/23	21	Alpha Carbon Chemistry: Enols and Enolates
11/6/23	21	Alpha Carbon Chemistry: Enols and Enolates
11/8/23	22	Amines
11/10/23	22	Amines
11/13/23	22	Amines
11/15/23	22	Amines
11/17/23		Exam 4 (Chapters 21 and 22)
11/20/23-11/24/23		No Class, Thanksgiving Break
11/27/23	24	Carbohydrates
11/29/23	24	Carbohydrates
12/1/23	24	Carbohydrates
12/4/23	25	Amino Acids, Peptides, and Proteins
12/6/23	25	Amino Acids, Peptides, and Proteins
12/8/23	25	Amino Acids, Peptides, and Proteins
12/11/23-12/14/23		Final Exam Time To Be Determined

THE FINAL EXAM TIME IS TO BE DETERMINED. DO NOT SCHEDULE VACATION TRAVEL UNTIL AFTER YOUR SCHEDULED EXAM.